

**ERRATA for**  
***PE Chemical Sample Questions and Solutions***  
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**Question 104, p. 9:**

The third sentence should read as follows:

The flow rate (lb/min) of the recycle is most nearly:

**Question 130, p. 25:**

The heat flux data should read as follows:

$$\text{Heat flux} \quad 650 \text{ Btu}/(\text{hr}\cdot\text{ft}^2)$$

**Solution 104, p. 56:**

Line 12 should read as follows:

$$X_{\text{KNO}_3 R} = 15.4 \text{ lb KNO}_3 / (15.4 \text{ lb KNO}_3 + 33.6 \text{ lb H}_2\text{O}) = 0.3141$$

**THE CORRECT ANSWER IS (B)**

**Solution 106, p. 57:**

The first sentence should read as follows:

First, determine how much fuel was burned to produce 100 lb mole of dry flue gas (CO and CO<sub>2</sub> only).

**Solution 135, p. 75:**

Line 2 should read as follows:

Notation:  $[ ]$  = activity and  $( )$  = mol/L

$$\begin{aligned} K_1 &= \frac{[\text{H}^+][\text{H}_2\text{PO}_4^-]}{[\text{H}_3\text{PO}_4]} = \frac{(\text{H}^+)\gamma_{\pm}(\text{H}_2\text{PO}_4^-)\gamma_{\pm}}{(\text{H}_3\text{PO}_4)\gamma_u} = 7.516 \times 10^{-3} \\ &= \frac{(x)0.8(x)0.8}{(1-x)1.0} = 7.516 \times 10^{-3} \end{aligned}$$

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**Solution 136, p. 76:**

The third line should read as follows:

That is, for  $\frac{4}{3} \text{ Al} + \text{O}_2 \leftrightarrow \frac{2}{3} \text{ Al}_2\text{O}_3$