Revisions are shown in red.

Question 51, p. 38:

The figure should appear as follows:
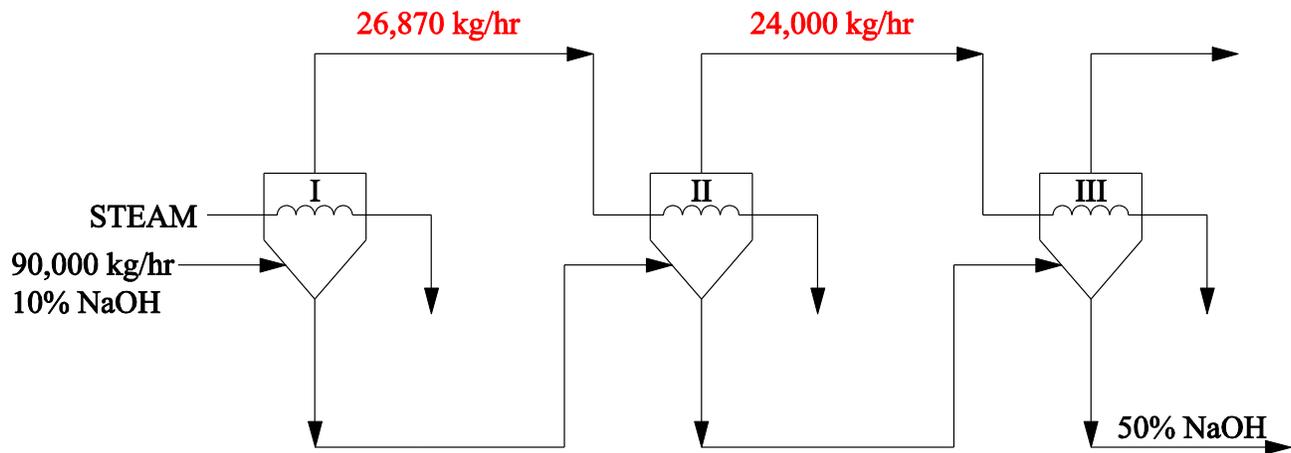
Solution 41, p. 89:

Line 4 should read as follows:

$$R = \frac{\overline{R}}{m} = \frac{8,314 \text{ J}}{\text{kmol} \cdot \text{K}} \frac{\text{kmol}}{28 \text{ kg}} = 297 \frac{\text{J}}{\text{kg} \cdot \text{K}}$$

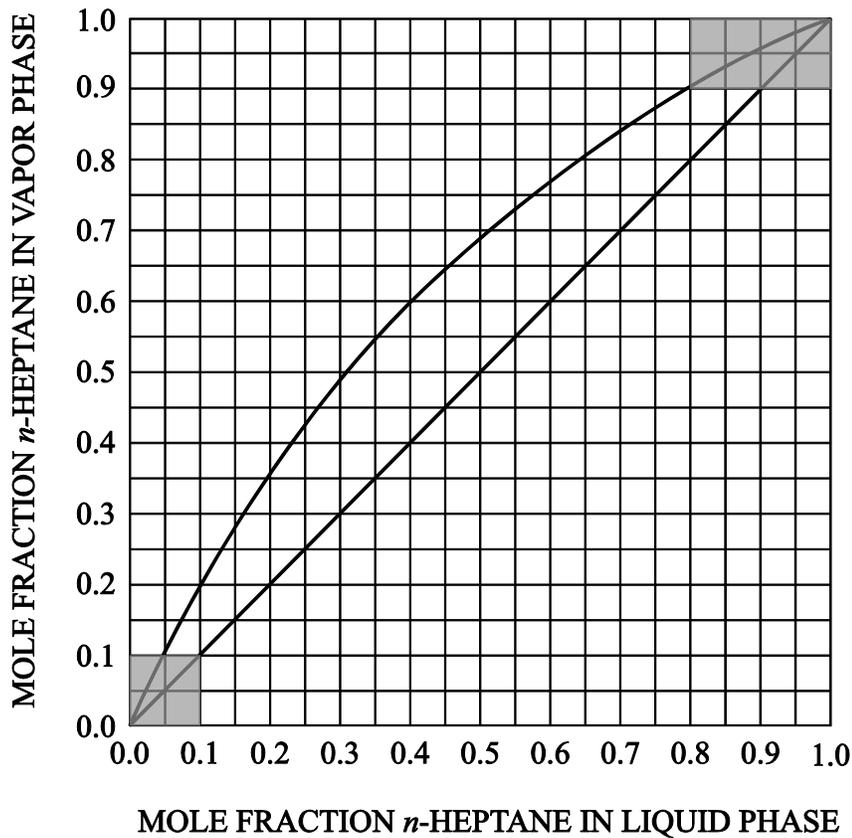**Solution 52, p. 95:**

Line 2 on p. 95 should read as follows:

$$\frac{826.7 - 2x}{1,230 - 3x}$$, but this must equal 0.7045.

Solution 71, p. 103:

The solution should read as follows:

Refer to the Murphree Plate Efficiency section in the Chemical Engineering chapter of the *FE Reference Handbook*.

CLICKS IN THE GRAY SHADED AREA WOULD BE SCORED AS CORRECT.

Solution 92, p. 111:

The last line of the solution should read as follows:

$$C/L = \frac{G_2}{1 + G_2 G_1 G_C H}$$

THE CORRECT ANSWER IS: A